

HW10 Solutions

Prove $E_F = \frac{\hbar^2 \pi^2}{2m} \left(\frac{3n_e}{\pi}\right)^{2/3}$.

Number of electrons energy between E & $E+dE$ are

$$N(E)dE = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} \frac{\sqrt{E} dE}{e^{\beta(E-E_F)} + 1}.$$

So, the total number of particles N is

$$N = \int N(E)dE = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} \int_0^\infty \frac{\sqrt{E} dE}{e^{\beta(E-E_F)} + 1}.$$

At $T=0$, it becomes $N = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} \int_0^{E_F} \sqrt{E} dE$.

Calculating this, we can get $N = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} \frac{2}{3} E_F^{3/2}$.

With $\frac{N}{V} = n_e$, we get $E_F = \frac{\hbar^2 \pi^2}{2m} \left(\frac{3n_e}{\pi}\right)^{2/3}$.

Chapter 11

Prob 14.

$$T_D = 225K$$

(a) $T=4K$, heat capacity $C = 0.0134 J/mol \cdot K$.

Lattice contribution to heat capacity

$$C_L = \frac{12\pi^4 R}{5} \left(\frac{T}{T_D}\right)^3 \simeq 0.0109 J/mol \cdot K$$

So, electronic contribution is $C_e = 0.0025 J/mol \cdot K$.

(b) $T=2K$

Lattice contribution to heat capacity $C_L \propto T^3$,

$$C_L \simeq 0.00136 J/mol \cdot K$$

Also, electronic contribution $C_e \propto T$,

electronic contribution is $C_e = 0.00125 J/mol \cdot K$.

So, total heat capacity is $C = 0.00261 J/mol \cdot K$.

Prob 15.

(a) $T = 293K \Rightarrow kT \simeq 0.02525 eV$

$$f_{FD}(E) = \frac{1}{e^{(E-E_F)/kT} + 1} = 0.1$$

$$\Rightarrow E = E_F + kT \ln(9)$$

$$\simeq 7.09 eV$$

$$(b) f_{FD}(E) = \frac{1}{e^{(E-E_F)/kT} + 1} = 0.9$$

$$\Rightarrow E = E_F - kT \ln(9)$$

$$\simeq 6.98 eV$$

So, energy range of $0.11 eV$.

Prob 17.

$$T = 293K \Rightarrow kT \simeq 0.02525 eV$$

$$\text{Eq (10.47) : } dN = V \frac{8\pi \sqrt{2} m^{3/2}}{h^3} E^{1/2} \frac{1}{e^{(E-E_F)/kT} + 1} dE$$

$$\text{implies } \frac{dN}{V} = \frac{8\pi \sqrt{2} m^{3/2}}{h^3} E^{1/2} \frac{1}{e^{(E-E_F)/kT} + 1} dE \simeq 2.3 \times 10^{24} m^{-3}$$

Prob 18.

$$\text{Eq (11.19) : } \sigma = \frac{ne^2\tau}{m}$$

$$\text{implies } \tau = \frac{\sigma m}{ne^2} \simeq 2.50 \times 10^{-14} s.$$

So, mean free path $l = \tau v_F \simeq 39.2 nm$,

which is 150 times larger than lattice spacing of copper ($0.256 nm$).

Prob 20.

$$\text{The current density } j = \frac{i}{A} \simeq 1.27 \times 10^4 A/m^2.$$

$$\text{So, } v_d = \frac{j}{ne} \simeq 9.37 \times 10^{-7} m/s.$$

With $dN = N(E)dE \simeq N(E_F)dE = N(E_F)mv_F v_d$ and

$$N(E_F) = \frac{3}{4} \frac{N}{E_F} \text{ (from eq (10.47) \& eq (10.49)),}$$

$$\text{we have } \frac{dN}{N} = \frac{3}{4} \frac{mv_F v_d}{E_F} \simeq 8.9 \times 10^{-13}.$$

Prob 22.

$$T = 293K \Rightarrow kT \simeq 0.02525 eV$$

$$E - E_F = \frac{E_g}{2} \text{ (by problem assumption)}$$

So,

$$f_{FD}(E) = \frac{1}{e^{(E-E_F)/kT} + 1} = \frac{1}{e^{E_g/2kT} + 1} \simeq 5.0 \times 10^{-48} \text{ for C}$$

$$f_{FD}(E) = \frac{1}{e^{(E-E_F)/kT} + 1} = \frac{1}{e^{E_g/2kT} + 1} \simeq 3.5 \times 10^{-10} \text{ for Si.}$$

The ratio of electron concentration is 1.4×10^{-38} .